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AND PHASE BEHAVIOR OF MIXTURES OF NITRIC OXIDE AND NITROGEN DIOXIDE

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Experimental measurements were made of the volumetric behavior of institutes of nitric oxide and nitrogen dioxide for pressures up to 7,000 provides per square inch in the temperature interval between 40° and 340° F.

South of these compositions were investigated in detail and the remainder more studied at only one specific volume. These measurements were made to constant volume equipment (1, 2) by following the change in pressure with temperature at a constant weight of sample. The total quantity of mixture to the isochoric equipment was varied in the case of four mixtures by withdrawing portions of the sample at a homogeneous state and determining the otherwise in weight of sample by weighing bomb techniques. Dew points and thinks points were established by discontinuities in the isochoric pressure-

The total volume of the spherical isochoric vessel used in these measurements was determined as a function of temperature and pressure from a browledge of the volume at one state and the thermal expansion and Young s medulus of the steel employed in its construction. The spherical form of the resel together with a nearly uniform wall thickness permitted such strain calculations to be made with accuracy. The volumetric characteristics

of the vessel were checked by calibration with water using the data of Smith and Keyes (3) for the volumetric behavior of the latter compound at elevated temperatures and pressures. Agreement with the calculated total volume of the sphere within 0.1% was obtained at temperatures between 40° and 340° F. and for pressures up to 7,000 pounds per square inch. In the course of the present work the volumetric characteristics based upon the properties of the steel were employed to determine the effect of temperature and pressure upon the total volume of the sphere.

Table I records corresponding values of pressure, temperature, and specific volume for each of the four mixtures which were investigated systematically. For each of these mixtures the pressure-temperature relations were determined for a number of different weights of sample. Table II records corresponding values of pressure, temperature, and specific volume for the 11 other mixtures investigated. In each of these cases only a single weight of sample was studied. The absence of a series of measurements at different weights of sample (different specific volumes) for a mixture of fixed composition results in difficult direct interpolation of the data to even values of pressure, temperature, and composition.

The pressures recorded in Tables I and II involve a standard error of 0.1% of the value or 0.3 pound per square inch, whichever is the larger measure of uncertainty. The weight of sample employed was known within 0.03% and the standard error in predicting the total volume of the

working vessel was 0.1%. It is believed that the specific volumes were established with a standard error of 0.15%. The temperature of the vessel was determined with a platinum resistance thermometer with a standard error of 0.03° F. from the international platinum scale. The estimates of small standard error were confirmed by the precision of the measurements which was about one-half of the total of the standard error of the several variables involved.

The results of the experimental measurements of the bubble point and dew point states are recorded in Table III for four mixtures which were investigated systematically and 11 mixtures which were studied at a single total weight of sample. These data were established from discontinuities in the nearly isochoric pressure-temperature derivative obtained from the data presented in Table 1. The uncertainty in detaxmining the bubble point and dew point states depends upon proximity to the critical state of the mixture. In the immediate vicinity of the critical state the change in slope of the isochoric pressure-temperature relation at the phase boundary is small. For this reason the uncertainty in the location is several times the standard error realized in determining experimentally the corresponding values of pressure and temperature. At a distance from the critical state the uncertainty in establishing the phase boundary is not more than twice the standard error of measurement.

REFERENCES

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- 3. Smith, L. B., and Keyes, F. G., Proc. Am. Acad. Arts and Sci., 69, 285 (1934).

TABLES

- I. Specific Volume of Four Mixtures of Nitric Oxide and Nitrogen Dioxide.
- II. Pressures and Temperatures for Mixtures of Nitric Oxide and Nitrogen Dioxide at Nearly Fixed Specific Volume.
- III. Experimental Dew Point and Bubble Point States.



TABLE I

SPECIFIC VOLUME OF FOUR MIXTURES OF NITRIC OXIDE AND NITROGEN DIOXIDE

Pressure,	Temp-	Specific	Pressure,	Temp-	Specific
Lb./sq. In.	erature	Volume,	Lb./ 5 q. i n.	erature	Volume,
Absolute	°F.	Cu. Ft. / Lb.	Absol ute	°F.	Cu. Ft./Lb.

Weight Fraction Nitrogen Dioxide = 0.94305

Sample Weight = 0. 792956 1b.		Sample Weight= 0.744286 lb.			
24. 19	40.07	0.011008	71. 73	110.01	0.011748
25.55	42.05	0.011009	77. 95	115.02	0.011749
25. 68	43.98	0.011009	81.45	117.99	0.011750
26.03	45 . 99	0.011010	81.68	118.01	0.011750
26.72	47.98	0.011010	166.33	119.03	0.011751
82.72	50.04	0.011011	265.31	120.00	0.011751
209.81	51,00	0.011011	1284. 26	130.01	0.011755
335, 21	52,05	0.011012	2305, 54	140.01	0.011760
1326.22	60,00	0.011015	3320.02	150.00	0.011764
2576,81	70.01	0.011020	4336.61	160.00	0.011768
3824, 14	80.05	0.011024	5358.91	170.02	0.011772
5044.04	90.01	0.011028	6909.73	185,00	0.011779
6263.92	100,02	0.011032			

Sample Weight= 0.782572 1b.

31.17	55.00	0.011158
31.73	55.00	0.011158
34.82	5 9.99	0.011160
32, 25	60.01	0.011160
36.63	64.00	0.011161
37. 08	64. 00	0.011161
149.18	66.00	0.011162
633.56	70.01	0.011163
1834.21	80.00	0.011167
1845.75	80.23	0.011167
3023.12	90.00	0.011172
4206,07	100.00	0.011176
5379.52	110.01	0.011180
7144, 18	125.00	0.011187

Pressure,	Temp-	Specific	Pressure,	Temp-	Specific
Lb./Sq. In.	erature	Volume,	Lb. /Sq. In.		
Absolute	°F.	Cu. Ft./Lb.	Abablute	°£`.	Cu. Ft./Lb.

Weight Fraction Nitrogen Dioxide = 0.94305

Sample Weight = 0.692236 1b.			Sample Weight = 0. 549798 1b.		
203.44	170.01	0.012650	47.65	80.00	C. 013892
220.73	175.00	0.012651	62.52	100.01	0. 01589 8
2 3 2.08	178.01	0.012652	145.69	150.00	0.015919
235.00	179.01	0.012652	384.94	210.07	0.015942
371.27	181.01	0.012653	701.60	250.00	0.015958
459.17	182.02	0.012654	932.13	270.02	0.015966
1133.93	. 190.02	0.012657	997.38	275.04	0.015968
1990.65	200.01	0.012661	1144.55	280.01	0.015970
2852.29	210.00	0.012666	1252.95	282.01	0.015971
2852, 38	210.00	0.012666	1698.59	255 02	0.015975
4164.37	224.99	0.012672	1698.85	290.02	0.015975
5517, 45	240.00	0.012679	2869.61	310.01	U. 015985
6536.41	251 . 02	0.012684	3802.52	325.01	0.015993
6910.41	255.05	0.012685	4778.50	340.02	0.016000
			4 782.89	340.00	0.016000

Sample Weight = 0.617948 1b.

126.08	140.02	0.014160
170.52	160.01	0.014167
325.80	200.01	0.014181
559.69	235.00	0.014193
602.62	240.00	0.014195
620.12	242.01	0.014195
677.80	243. OC	0.014196
742.75	2 44 , 00	0.014196
1152.68	250.00	0.014199
2200.13	265.02	0.014206
3293.27	280, û2	0.014213
3293.46	280.01	0,014213
4434, 74	295.03	0.014220
5612.05	310.02	0.014227
6841.76	325.03	0.014234

Pressure,	Temp-	Specific	Pressure,	Temp-	Specific
Lb./Sq. In.	erature	Volume,	Lb./Sq. In.	erature	Volume,
Absolute	°F.	Cu. Ft./Lb.	Absolute	°F.	Cu. Ft. /Lb.

Weight Fraction Nitrogen Dioxide = 0, 94305

Sample	Weight = 0.4	460197 1 6.	Sample i	Veight-0.2070 8	16 1b.
445.03	220.65	0.019051	679.70	2 49.9 8	0.042368
9 91.98	275.02	0.019077	1163.80	289.99	0.042410
1136.97	285.04	0.019082	1432.66	304.9 9	0.042427
1218.97	290.04	0.019085	1529.01	309. 99	0.042433
1305.01	295.03	0.019087	1589.2 8	313. 90	0.042437
1397.10	300.04	0.019090	1658.07	318.02	0.042442
1518.41	305.04	0.019092	1720.58	321.99	0.042446
1715.51	310.04	0.019095	178 2,88	326.00	0.042450
2124, 27	320.04	0.019100	1846.07	329, 97	0.042454
2554.30	330 . 06	0.019106	2008.29	339 , 99	0.042465
2994.89	340,28	0.019111	2024.19	340. 02	0.042465

Sample Weight s 0. 356587 lb. 451, 51 220.00 0.024586 691.55 250.01 0.024605 1048.46 280.00 0.024623 0.024636 1380.19 300.01 1577.78 310.03 0.024643 1692.83 315.01 0.024646 1806.96 319.26 0.024649 330.00 2104.77 0.024656 0.024663 2404.30 340.00 2412.77 340.03 0.024663

Pressure,	Temp-	Specific	Fressure,	T'emp -	Specific
Lo./Sq. in.		Volume, Cu. Ft./Lb.	Lb./Sq. In. Absolute		Volume, Cu. Ft./Lb.
110001416	4 1	Q 31 I 11 / LD1	. ID BOILEC	• •	Qu, 11, 150,

Weight Fraction Nitrogen Dioxide - 0,82046

Sample	Sample Weight: 0.745091 ib.		Sample Weight +0.698354 1b.		
31.94	60.00	0.011721	178.43	140.01	0.012530
73.65	95.00	0.011731	215.59	150.01	0.012533
81.16	100.00	0.011732	257.90	159.99	0.012536
545, 65	105.00	0.011734	403.12	162.03	0.012537
1093.55	110.00	0.011737	68 3.63	165. 0 0	0.012538
2209.09	120.00	0.011741	1157.93	170.00	0.012540
3327.09	130.00	0.011745	1628.69	175.00	0.012542
4436.51	140.00	0.011750	3076, 80	189.99	0.012549
4438.04	140.01	0.011750	4537.82	205.01	0.012556
5543, 97	150.00	0.011754	6035.13	220.00	0.012562
6665.61	159.99	0.011758	7042.21	229.99	0.012567

Sample Weight = 0, 723010 lb.

110.00	0.012094
120.00	0.012097
130.00	0.012100
131.99	0.012100
135,00	0.012100
140.00	0.012104
150.01	0.012108
160.00	0.012113
175.0C	0.012119
189.99	0.012126
	120.00 130.00 131.99 135.00 140.00 150.01 160.00 175.00

Pressure,	Temp-	Specific	Pressure,	Temp-	Specific
Lb./Sq. In.	erature	Volume,	Lb./Sq. In.	erature	Volume,
Absolute	Ĕŗ.	Cu. Ft./Lb.	Absolute	°F.	Cu. Ft./Lb.

Weight Fraction Nitrogen Dioxide = 0, 82046

Sample	• Weight = 0.667990 1b.		sample Weight = 0, 667990 lb. Sample Weight = 0, 551141 lb			1141 15.
308. 23	170.02	0.013109	5 0.69	50.00	0.015953	
3 63.08	180.00	0.013112	322.08	235.01	0.015914	
427.98	190.00	0.013115	1013.69	250.01	0.015920	
639.83	195,00	0.013117	1163.03	260.01	0.015924	
1071.71	200.00	0.013119	1241.32	265.01	0.015926	
1507.85	205.01	0.013122	1242.55	265.00	0.015926	
2827.12	219.76	0.013128	1337.46	270.01	0.015928	
4202.02	234.98	0,013135	1390.30	271.76	0.015929	
5608, 78	250.00	0.013142	1602.26	275, 24	0.015930	
6559.05	250.01	0.013147	1895, 22	28A. 00	0.015933	
			2866.55	295.01	0.015940	
			3881.29	310.01	0.015948	
			4942.01	325.01	0.015956	
			6044.10	340.00	0.015964	

Sample Weight = 0.632528 1b.

488.88	200.00	0.013854
571.05	210.01	0.013858
664.76	220.00	0.013861
813.62	225.03	0.013863
1228.21	230.23	0.013865
1596.61	234.99	0.013867
2821.09	~250.01	0.013874
4093.08	265.00	0.013882
5394.79	280.00	0.013889
6739.01	295.00	0.013896

Pressure, Lb./Sq. In. Absolute	erature	Specific Volume, Cu. Ft./Lb.	Pressure, Lb./Sq. in. Absolute	Temp- erature °F.	Specific Volume, Cu. Ft./Lb.
	Wei	ght Fraction Nit	rogen Dioxide = 0	, 94059	-
Sample W	eight = 0. 1	98 594 16.	Sample We	ight = 0.086	3081 15.
45.08 150.68 417.11 1013.57 1204.48 1333.23 1427.00 1521.00 1673.96 1977.93	100.00 159.99 220.00 280.00 292.55 300.00 305.01 310.00 320.03 340.00	0.043989 0.044052 0.044117 0.044183 0.044217 0.044229 0.044206 0.044212 0.044197 0.044251	396. 79 624. 80 776. 64 833. 26 924. 14 1041. 70 1159. 31 1274. 68 1276. 15	220.00 250.01 264.99 270.00 280.00 249.99 320.00 340.01 340.00	0. 10151 0. 10163 0. 10164 0. 10166 0. 10171 0. 10176 0. 10181 0. 10181
Sample V	reight=0.0	18907 15.	Sample We	eight = 0, 01:	1429 lb.
195.47 374.70 405.69 447.20 482.79 556.17 628.03 697.00	180.00 220.00 225.00 235.00 250.00 280.00 310.00 340.00	0. 22497 0. 22519 0. 22534 0. 22537 0. 22534 0. 22551 0. 22568 0. 22568 0. 22584	41.03 105.97 127.06 138.89 145.31 153.7) 185.16 231.17 271.87	100.00 150.00 160.00 165.00 170.00 180.00 220.00 280.00 340.00	0.65052 0.65241 0.65146 0.65154 0.65162 0.65178 0.65241 0.65334 0.65428

Pressure,	Temp-	Specific	Pressure,	Temp-	Specific
Lb./Sq. In.	erature	Volume,	Lb./Sq. In.	erature	Volume,
Absolute	°F.	Cu. Ft./Lb.	Ab sol ute	°F.	Cu. Ft./Lb.
	₩e	ight Fraction Ni	trogen Dioxide = 0	. 83125	
Sample \	Weight = 0.	244.25 16.	Sample W	eight =0, 10	9642 1b.
75.84	100.01	0.035784	80 4. 11	249.97	0.074910
23 2.86	160,00	0.035837	916.39	259.99	0.079990
5 93.92	220.00	0.0 35 89 0	1041,58	269.99	
	279.99	0.035944	1182.32	280.00	0.080031
1510.82	290.00	0.035953	1275. 25	290.00	
1607.99	295.00	0.035958	1432.98	310.00	
	30 0.00	0. 035 96 3	1668.50	340.01	0.080150
•	305.00	0. 035 96გ			
1924.51	\$10.00	0.035972			
2027.42	315.00	0.035977			
2126.22	320.00	0. 035 982			
2535.54	340,00	0.036001			
Sample \	Weight= 0.	055316 1ь.	Sample W	eight=0.02	4337 lb.
416.64	210.00	0.15835	169.88	160.00	0. 35 888
550.47	229.99	0.15843	231.25	180.00	0.35906
630.17	240,01	0.15847	269. 16	190.00	0.35915
692.08	250,00	0.15851	302.96	200.00	0.35923
800.22	280.00	0.15862	331.45	219.76	0.35941
906.8 3	310.00	0.15874	376.31	250.01	0.35967
1012.37	340.00	0.15885	463.48	310.00	0.36018
			300.35	340.00	0.36044
Sample V	Veight 0.0	006350 1ь.			
39, 27	88.40	1. 3753			
47. 16	99.98	1.3757			
56.37	110.00	1.3760			
67.70	130.00	1.3764			
72.58	130.00	1. 2770			
76.20	140.90	1.3770			
83.26	160.00	1.3777			
106.55	220.00	1. 37 97			
124.86	285.00	1.3816			
141.74	340.00	1.3836			
- · · - • • •	 -	= + - * * *			

TABLE II

PRESSURES AND TEMPERATURES FOR MIXTURES OF NITRIC CXIDE AND NITROGEN DIOXIDE AT NEARLY FIXED SPECIFIC VOLUME

Pressure, Lb./Sq. in. Absolute	Temp- erature or,	Specific Volume, Cu. Ft. / Lb.	Pressure, Lb./Sq. In. Absolute	_	Specific Volume, Cu. Ft. / Lb.			
	Nitrogen Dioxide							
	Weight Frac	—	Weight Fra	ction 0,883	336			
	Sample Wei	ght=0,034363 lb.	Sample Wel	lght = 0. 4566	2 16.			
52, 42	100.00	0.25668	26.65	70.00	0.19268			
52.49	100.00	0.25668	52.31	100.00	0.19283			
159, 11	160.00	0.25709	63.20	100.00	0.19283			
349.96	210.00	0.25742	96 .06	130.00	0.19299			
397.53	220.00	0.25748	164.10	160.00	0.19313			
424.72	230.00	0.25756	267.46	190.00	0.19328			
469, 04	250.00	0.25769	421.31	220.00	0.19344			
534, 50	280.00	0.25790	486.01	230.00	0.19348			
598.84	310.00	0.258 09	542.89	240.00	0.19354			
			575.67	250.00	0.19359			
			665, 16	280.00	U. 19374			
			755.4 8	310.00	0.19389			
			841.05	340.00	0.19404			
			843.15	340.00	0.19404			
			920.68	370.00	0.19419			
			9 91. 58	400.00	0.19434			
		Nitrogen Dio	zide					
	Weight Frac	tion = 0.88114	Weight Frac	tion = 0.791	31			
	Sample Weig	ht =0.023025 lb.	Sample Weight = 0. 028264 1b.					
12.86	40.00	0.38182	19.20	40.00	0.34281			
50,45	100.00	0.38241	69.60	100.00	0.34335			
5 0. 79	100.00	0.38243	70.3 8	100.00	0.34335			
151, 17	160.00	0.38301	188.85	160.00	0.34389			
245.55	190.00	0.38331	253.75	180.00	0.34407			
273.13	260.00	0.38341	292.80	190.00	0.34416			
2 87 . 82	210.00	0.38351	329.05	200.00	0.34424			
301, 74	220.00	0.38361	361.92	220.00	0.34443			
3 44 . 25	250.00	C. 38391	408.85	250.00	0.34470			
386. 54	280.00	0.38421	455, 79	280.00	0 .344 96			

Pressure, Lb./Sq. In. Absolute	-	Specific Volume, Cu. Ft. / Lb.	Pressure, Lb./Sq. in. Absolute	-	Specific Volume, Cu. Ft. / Lb.
		Nitrogen Diox		/ 1	
	Weight Fra	ction = 0.78280	Weight Fract	ion =), 61/7	U
	Sample We	ight + 0.043282 1b.	Sample Weig	ht #C.03278	6 1 5.
22.31	40.00	0.22662	51, 1 3	4C.00	0.26814
24.22	40.00	0.22662	52, 74	40.00	C. 26 814
79.35	100.00	0.22698	1 35.4 0	100.00	0.2 6\$ 5 6
79.57	100.00	0.22698	1 36.0 0	100,00	0.26 \$ 5 6
81.14	100.00	0.22698	29 3 . 84		0.2 68 9 8
213.96	160.00	0.22733	373.1 8		0. 2 69 1 2
43 2.01	210.00	0.22763	419.73	19(,00	0.2692 0
489.94	220.00	0.22769	471, 21	20(.0 0	0.26927
524.86	230.00	0.22775	504. 0 8	2 1 0.0 0	0.26934
574, 95	250.00	0.22787	5 25.0 0		
648.80	280.00	0.22805	585 . 5 9	250.00	0. 2 6962
721.43	3 10.00	0.22822	6 44.8 4	280.00	C. 2 6983
		4414	1.4.		
	119 - 1 - 4 - 4 - 4	Nitrogen Diox		U A 2711	1.7
	Weight FT	action =0.39620	Meight Figur	non=U.Z/I:	
	Sample We	eight =0,051115 lb.	Sample Weig	tht = (. 0745	74 1b.
3 68. 11	40.00	0.17199	820 . 7 5	4 (.0 0	0.11789
3 68. 97	40.00	0.17199	925 . 5 8	10(,00	C. 1180:
380,50	40.00	0.17199	95 3.3 3	10(,00	0.11805
447.36	100.00	0.17226	1025.92	13(.00	C, 1181?
450,27	100.00	0.17226	11 34.5 8	16(.00	C. 1182
519.39	139.00	0.17240	1285.21	19(.0 0	C.1183>
519.98	130.00	0.17240	1344.31	2 0 (.0 0	C. 1183;
623.91	160.00	0.17253	1351.47	20(.00	0.11830
717.69	180.00	0.17262	1 37 9. 7 9	201,00	0.11841
772.98	190.00	0.17267	1421.24	210,00	0,1 18 4 2
834.76	200.00	0.17271	1479.73	22(.00	C.1184 3
903.02	210.00	0.17276	1616.90	25(.00	0.11 85 <i>3</i>
942.41	220.00	0.17281	1662.22	261,00	0.1185
1038, 27	250.00	0.17294	1752.62	28(.00	0.11864
1132, 57	280.00	0.17308			

4	-		_	_	
Pressure,	Temp-	Specific	Pressure,	Temp-	Specific
Lb./Sq. In.	erature	Volume,	Lb./Sq. In.		Volume,
Absolute	°F.	Cu. Ft. / Lb.	Absolute	°F.	Cu. Ft. / Lb.
		Nitrogen Die	oxide		
Weight	Fraction= 0.	94508	Weight Fract	lon=0.7999	94
Sample	Weight=0.0	94347 16.	Sample Weig	ht = 0.66866	5 41 b.
21.45	73.71	0.357897	17.58	40.00	0.013055
39.62	100.00	0.358095	40.65	70.00	0.013064
73.81	130.00	0. 3 58 386	84.37	100.00	0.013073
130.05	160.00	0.358680	104.79	110.00	0.013077
219.32	190.00	0.358975	130.39	130.00	0.013083
257 , 67	200.00	0.359074	231. 85	150.00	0.013089
271.85	205.00	0.359124	274, 13	160.00	0.013093
287 , 08	215.00	0. 359221	386.01	180.00	0.013099
294.09	220.00	0.359270	421.00	184.99	0.013102
319.72	250.00	0.359562	465, 14	190.00	0.013102
385, 26	280.00	0. 359855	6 39.4 0	191.98	0.013103
429.58	310.00	0.360147	907, 53	195.00	0.013104
470.37	340.00	0.360449	1348.04	199.99	0.013167
			1796.61	208.00	0.013109
			3159.33	220.00	0.013100
			5022.68	240.00	0.013100
			6951, 17	259.99	0.013100

Nitrogen Dioxide

Weight Fraction: 0,93755

Sample Weight = 0. 939622 1b.

37. 16	100.00	0.942166
68.21	130.00	0.942933
83.13	140.00	0.943189
96.63	150.00	0.943446
101.71	160.00	0.943700
117.99	190.00	0.9 44464
135. 19	220.00	0.945229
152.47	250.00	0. 945993
168. 29	280.00	0.946757
182.20	310.00	0.947521
194.66	3 40 ,00	<i>ù.</i> 9482 84

TABLE III

EXPERIMENTAL DEW POINT AND BUBBLE POINT STATES

-BUBBLE POINT

Pressure, Lb./Sq. Inch. Absolute	Temperature F.	Specific Volume, Cu. Ft. / Lb.	Pressure, Lb./Sq. Inch Absolute	Temperature or.
	Weight Fraction	Nitrogen Dioxid	e=0.9 430 5	
29	49.3	0.011011	50	85. 2
3 8	65, 2	0.011161	100	129.6
8 2	118.0	0.011750	150	152.1
235	179.2	0.012653	200	169. 3
62 3	242.0	0.014195	250	182 .9
1029	277. 7	0.015970	300	194. 9
1466	3 03. 6	0.019091	400	213.6
1611	311.6	0. 024644	500	227.9
1540	310.5	0.042433	600	239.7
			800	259. 2
			1000	275.6
	Weight Fraction	n Nitrogen Dioxid	ie=0.82 04 6	
81	100.3	0.011732	50	74. 1ª
153	130.5	0.012100	100	110.3
258	160. 1	0,012536	150	129.7
445	192. 1	0.013116	200	145. 2
696	223, 1	0.013862	250	157.9
1334	270,4	0.015928	300	168.7
		***************************************	400	185.8
-			500	199.9
			600	212.6
			800	232.8
			1000	248. 1
			1250	265. 1

Weight Fraction Nitrogen Dioxides 0.79994

439 190 0.013102

TABLE III (cont.) DEW POINT

		DEW POINT		
Pressure, Lb. / Sq. Inch	Temperature oF.	Sp ecifi c Volume,	Pressure, Lb./Sq. Inch	Temperature F.
Absolute	(Du. Ft. / Lb.	Absolute	
	Weight Fractio	n Nitrogen Dioxide-	~0. 94 059	
143	166.6	0,65157	50	116.4 ^a
433	429.6	0,22524	10C	146, 9ª
8 96	275.1	0.10165	130	169.0
1498	308.4	0.044215	200	185, 5
			250	1 9 6. 9
			300	208.0
			400	224.9
			500	239.0
	•		600	250.3
			800	268.0
			1000	282,4
			1250	296.9
			1500	308.6
	Weight Fraction	n Nitr ogen Dioxide	0.83125	
69	121.3	1, 3764	50	105.3ª
2 9 9	197.7	0.35941	100	139.9
677	246.1	0.13849	150	160.4
1221	283. 0	0.08 0037	200	175.0
1984	3 12.9	0.035975	250	187.0
			30 0	197.9
			400	215.3
- .			500	228.9
			600	239 2
			800	256.1
			1000	269.9
			1250	284.4

DEW POINT

Weight Fraction Nitrogen Dioxide	Pressurt, Lb. / Sq. Inch Absolute	Temperature F.	Specific Volume, Cu. Ft./ Lb.
0.88527	405	221	0.25750
0.88 33 6	531	23 6	0.19352
0.881 33	26 3	193	0.38334
0.79131	325	197	0.31232
0.78280	30 8	223	0.20409
0.61770	491	204	0, 26929
0.39620	918	214	0.17278
0.27157	1470	219	0,1185
0.94508	268	202	0.33910
0.93755	94	147	0.94340

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